

Dimethyl Sulfide

CH₃-S-CH₃

CAS No: 75-18-3

Steam Cracker Applications

Dimethyl Sulfide (DMS) is a presulfiding agent for steam cracking, the process that turns hydrocarbons into ethylene and propylene used for olefin-based products. DMS is added to the feedstock to reduce coke and carbon monoxide formation during the high temperature thermal cracking of hydrocarbons in the tubular reactors. The deposition of coke on the boundary layer can significantly affect the run length, yields, and the coil life of a pyrolysis furnace. In addition, carbon monoxide is a poison for downstream catalysts. DMS acts as an efficient H₂S precursor, which allows for H₂S to form at the metal interface. The sulfur is absorbed to inhibit reactions.

Packaging

DMS is available in bulk (railcars, tank trucks and ISO containers) or packaged containers including 250 gallon returnable steel totes, 57 gallon returnable steel cylinders or 54 gallon drums.

Advantages

DMS directly and completely decomposes to H₂S at 1470°F (although some trace CS₂ may remain), suitable for furnace operating temperatures. See Figure 1 for the decomposition of DMS as well as DMDS which can also be utilized and begins decomposition to DMS and Methyl Mercaptan at a lower temperature. DMS usually requires only one injection point in the charge gas as DMS will disperse fairly well throughout the furnace system.

Safety and Handling

DMS is a highly flammable material and should be handled accordingly—stored under inert conditions and away from potential ignition sources. DMS is a dermal, oral, and toxic inhalation material with a highly unpleasant odor. It is slightly soluble with

Sulfiding Agent Properties			
Characteristics	DMDS	TBPS454	DMS
Sulfur %	68	54	52
Density (lbs/gal)	8.9	9.0	7.11
Freezing Point (°F)	-121	-54	-145
Boiling Point (°F)	229	291	99
Flash Point (°F)	59	217	-36
Vapor Pressure at 70°F (PSI)	0.45	<0.1	8.1
Decomposition Temperature* (°F)	1150	625	1470
Viscosity at 70°F (mPa.s)	0.62	12.8	.285

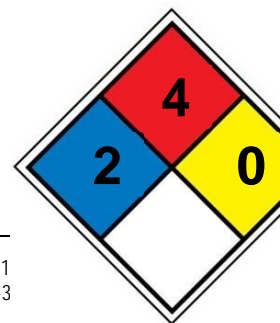
*In the presence of steam

compatible with both carbon and stainless steel as long as excess water is not present. Brass and copper connections are not recommended. Gaskets should be Teflon or Viton since DMS tends to attack rubber and nitrile elastomers. Household bleach (not pure bleach) or Liquid Alive® is suggested for any necessary clean up. When handling DMS, precautions should be made as pressure can build up in confined spaces such as hoses. Please reference the Material Safety Data Sheets for additional handling and safety recommendations.

Product Safety Information

Material Safety Data Sheets are available upon request and on our website:

www.cpchem.com/specialtychemicals



Application Guidelines

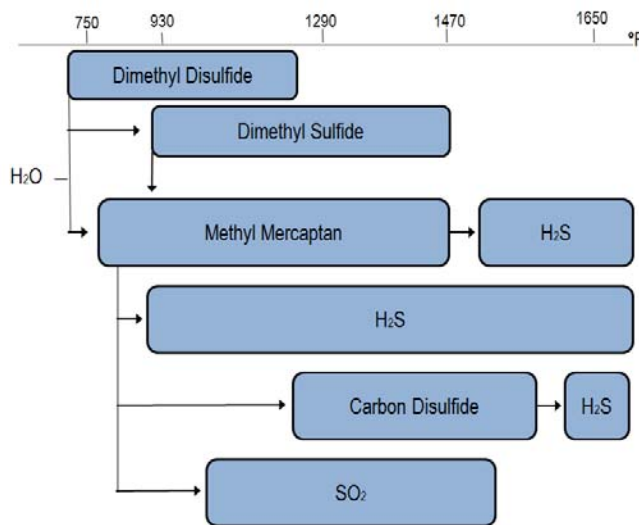
After furnace tubes are de-coked, feedstock with DMS is injected in order to passivate the furnace tubes. The rate of continuous injection is determined by the feed rate of the feedstock and its sulfur content. The lower the sulfur content of the feedstock, the higher the injection rate of the DMS needed. During the initial passivation step, DMS is normally injected at twice the normal rates. The feedstock is then mixed with steam and heated in a furnace for cracking. The radiant coils in the furnace are the predominant location where coking will occur. The gases are then sent to the transfer line exchanger, the second place where coking will occur. See figure 2 for basic steam cracker process. DMS, by decomposing to H_2S , will mitigate the creation of coke. The sulfur absorption blocks the peripheral site and prevents further coking by creating an iron sulfide layer at the reaction site. In addition, the H_2S reacts with the CO, formed by the reaction of hydrocarbons or coke with steam, by catalyzing the water gas shift reaction which converts the CO to carbon dioxide (CO_2). After the initial passivation step is complete, the DMS injection rate is reduced but continued to maintain a partial presence of H_2S in the furnace, maintaining the iron sulfide layer. Keep in mind that not all the sulfur content of the sulfiding agent converts to H_2S , as there is also SO_2 formation to be managed.

Other sulfur chemicals

In addition to DMS, DMDS and TBPS 454 can also be used for this application. Please reference those fact sheets for additional details.

Before using this product, the user is advised and cautioned to make its own determination and assessment of the safety and suitability of the product for the product for the specific use in question and is further advised against relying on the information contained herein as it may relate to any specific use or application. It is the ultimate responsibility of the user to ensure that the product is suited and the information is applicable to the user's specific application. Chevron Phillips Chemical Company LP does not make, and expressly disclaims, all warranties, including warranties of merchantability or fitness for a particular purpose, regardless of whether oral or written, express or implied, or allegedly arising from any usage of any trade or from any course of dealing in connection with the use of the information contained herein or the product itself. The user expressly assumes all risk and liability, whether based in contract, tort or otherwise, in connection with the use of the information contained herein or the product itself. Further, information contained herein is given without reference to any intellectual property issues, as well as federal, state or local laws which may be encountered in the use thereof. Such questions should be investigated by the user.

Figure 1. Decomposition of DMDS and DMS in the presence of steam



Actual results may vary depending on unit condition and pressures

Figure 2. Basic steam cracker process

